1) Approximate the direction of the net dipole of the following molecules

- a) O b) c) O
- 2) Label all non-hydrogen atoms sp1, sp2, or sp3
- c) II
- 3) Allenes are not flat but has 3-D geometry to it. Explain this using molecular orbitals.



- 4) Circle all stereocenters and label them as (R) or (S). Identify the molecules as chiral, achiral, or meso.
- a) b) O c) O d)
- 5) Convert the following Fisher projections into Sawhorse Diagrams or visa versa.
 - a) CH_3 b) NH_2 c) OH d) $HO CH_3$ HO CH_3 HO CH_3 OH
- 6) Convert the following into their lowest energy chair conformation.
 - a) b) i-Pr c) HO...
- 7) Draw all Eclipsed and Staggered Newman prjections of the following molecules and predict the highest and lowest energy.